

# Photochemistry: rates, opacity, and heating

processes overview and basics

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better science through chemistry

# Aims and goals



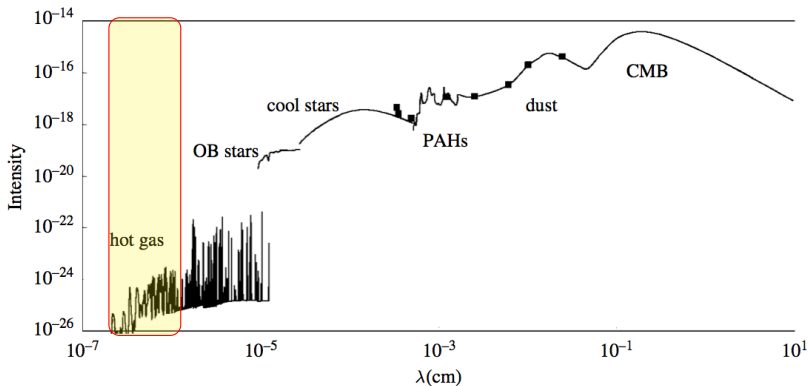
- ▶ have an overview of the radiation sources in the Universe
- ▶ how the radiation interact with the gas
- ▶ how the photochemistry works
- ▶ how much photodissociation is complex
- ▶ something about the optical depth

# Radiation sources in the Universe

mean intensity in units of  $\text{erg cm}^{-2} \text{s}^{-1} \text{Hz}^{-1} \text{sr}^{-1}$



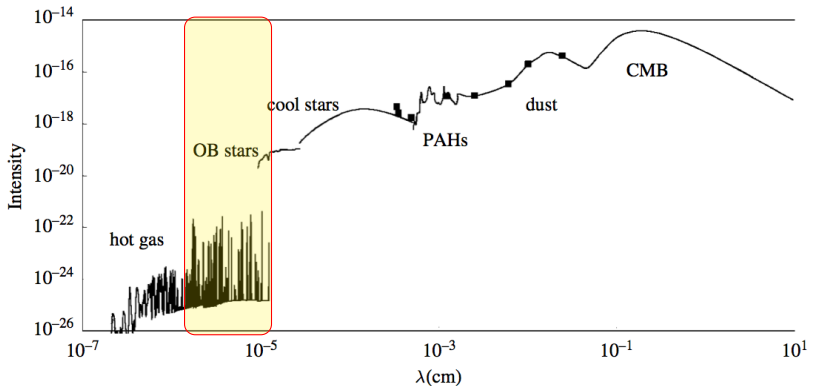
The ISM is permeated by various photon fields, which influence the physical and chemical state of the gas and dust.



► Diffuse X-rays ( $> 124 \text{ eV}$ )

# Radiation sources in the Universe

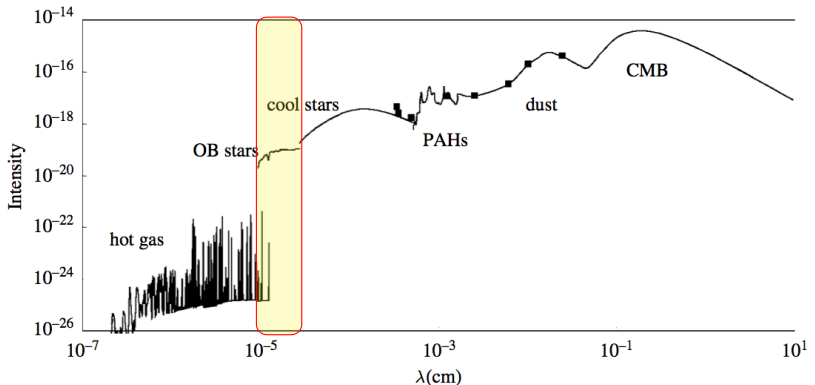
mean intensity in units of  $\text{erg cm}^{-2} \text{s}^{-1} \text{Hz}^{-1} \text{sr}^{-1}$



► EUV (10.25 – 124 eV)

# Radiation sources in the Universe

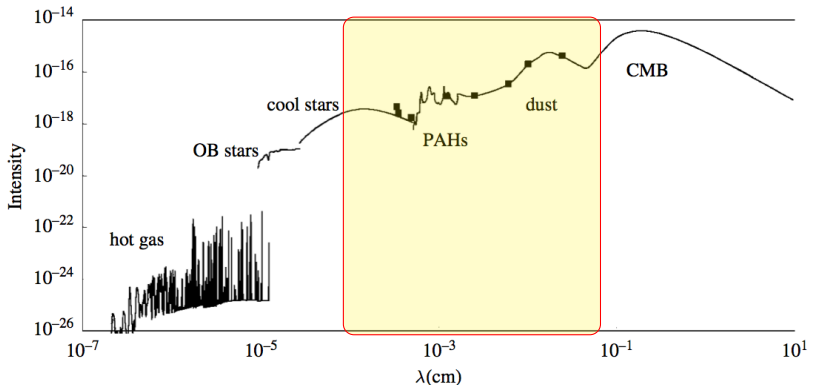
mean intensity in units of  $\text{erg cm}^{-2} \text{s}^{-1} \text{Hz}^{-1} \text{sr}^{-1}$



► FUV (6.20 – 13.6 eV, including Ly- $\alpha$  edge)

# Radiation sources in the Universe

mean intensity in units of  $\text{erg cm}^{-2} \text{s}^{-1} \text{Hz}^{-1} \text{sr}^{-1}$



► IR (0.0012 – 1.2 eV)

# Radiation sources in the Universe (cont'd)



- ▶ FUV emission mostly from early-type stars
- ▶ A-type stars visible region
- ▶ late-type stars far-red to near-IR
- ▶ hot plasmas (e.g. SNRs) X-rays
- ▶ Ly- $\alpha$  edge at 13.6 eV  $\rightarrow$  due to HI absorption

## Radiation needs to be included

- ▶ HI/HII regions, PDR
- ▶ shocks
- ▶ molecular clouds
- ▶ black holes
- ▶ ISM
- ▶ ...

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Reference: Tielens, 2005

# Main processes induced by radiation

overview



In any astrophysical region where UV photons can penetrate the following processes can occur:

1. photoionization
2. photodissociation
3. heating of the medium
4. (secondary ionizations)

Photoionization/photodissociation are the main destruction processes for atoms/molecules.



# Main processes induced by radiation (1)

photoionization



Photons with energies  $h\nu > E_0$  eject electrons from the atoms

## Atoms



- ▶  $n = 0 \rightarrow$  ionization of a neutral atom
- ▶  $E_0 \rightarrow$  ionization potential of the atom (see table)
- ▶ the electrons of energy  $(h\nu - E_0)$  produced can
  - ▶ elastically collide with ambient atoms/electrons (**heating sources**)
  - ▶ excite other ions/atoms (e.g. secondary ionization processes)

## Molecules



### Ionization potential

| Atom | $E_0$ (eV) | Atom | $E_0$ (eV) |
|------|------------|------|------------|
| HI   | 13.6       | OI   | 13.61      |
| Hel  | 24.6       | OII  | 35.1       |
| Hell | 54.4       | Sil  | 8.1        |
| CI   | 11.2       | Sill | 16.3       |
| CII  | 24.4       | Fel  | 7.9        |

# Photoionization

how to evaluate the rates



assume here an optically thin medium (see next slides)

$$k_{ph} = 4\pi \int_{\nu_0}^{\infty} \frac{I(\nu)\sigma(\nu)}{h\nu} d\nu \quad \Rightarrow \text{in terms of frequency}$$

$$k_{ph} = \frac{4\pi}{h} \int_{E_0}^{\infty} \frac{I(E)\sigma(E)}{E} dE \quad \Rightarrow \text{in terms of energy}$$

Note: **KROME** works in terms of energy

# Photoionization

how to evaluate the rates



- ▶ radiation flux per energy

$$k_{ph} = \frac{4\pi}{h} \int_{E_0}^{\infty} \frac{I(E)}{E} \sigma(E) dE$$

The term  $\frac{I(E)}{E}$  is highlighted in a blue circle, and  $\sigma(E)$  is highlighted in a red oval. An arrow points from the text "radiation flux per energy" to the blue circle.

---

Hz = s<sup>-1</sup>

# Photoionization

how to evaluate the rates



- ▶ radiation flux per energy

$$k_{ph} = \frac{4\pi}{h} \int_{E_0}^{\infty} \frac{I(E)}{E} \sigma(E) dE$$

- ▶ photoionization cross-sections

---

Hz = s<sup>-1</sup>

# Photoionization

how to evaluate the rates



- ▶ radiation flux per energy

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The equation is annotated with a blue circle around  $\frac{I(E)}{E}$  and a red oval around  $\sigma(E)$ . Arrows point from the text 'radiation flux per energy' to the blue circle and from 'photoionization cross-sections' to the red oval.

- ▶ photoionization cross-sections

- ▶ units check:  $\frac{\text{sr}}{\text{eV s}} \frac{\text{eVs}^{-1}\text{cm}^{-2}\text{Hz}^{-1}\text{sr}^{-1}}{\text{eV}} \text{cm}^2\text{eV}$

---

$$\text{Hz} = \text{s}^{-1}$$

# Photoionization

how to evaluate the rates



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- ▶  $\frac{\cancel{\text{sr}}}{\text{eV s}} \frac{\text{eVs}^{-1}\text{cm}^{-2}\text{Hz}^{-1}\cancel{\text{sr}^{-1}}}{\text{eV}} \text{cm}^2\text{eV}$

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# Photoionization

how to evaluate the rates



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# Photoionization

how to evaluate the rates



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$$k_{ph} = \frac{4\pi}{h} \int_{E_0}^{\infty} \frac{I(E)}{E} \sigma(E) dE$$

- ▶ photoionization cross-sections

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- ▶  $\frac{\text{sr}}{\text{eV s}} \frac{\text{eVs}^{-1} \text{cm}^{-2} \text{Hz}^{-1} \text{sr}^{-1}}{\text{eV}} \text{cm}^2 \text{eV} \Rightarrow [1/\text{s}]$

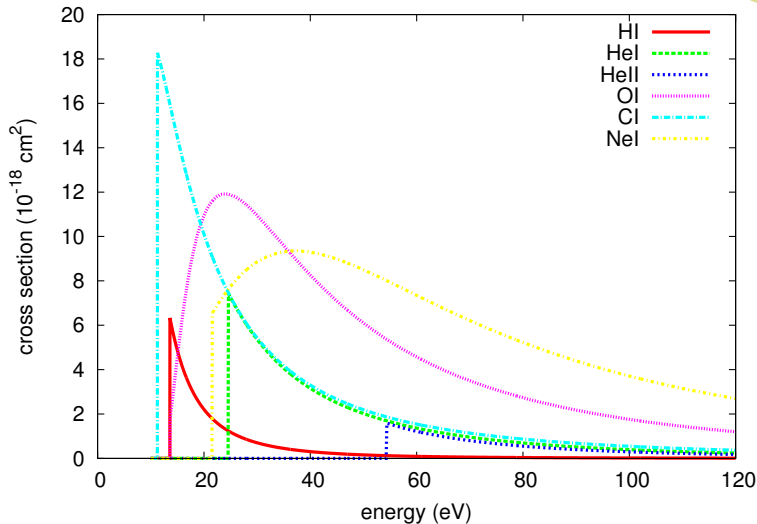
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$$\text{Hz} = \text{s}^{-1}$$



# Photoionization (cont'd)

the cross-sections



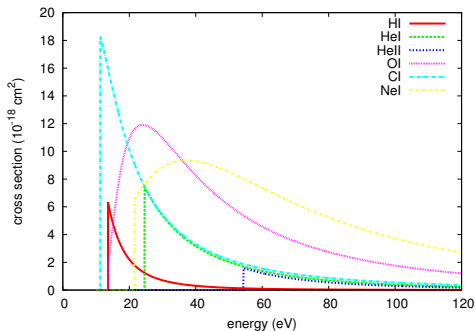
# Photoionization (cont'd)

the cross-sections



some considerations:

- ▶  $\sigma_H \propto \nu^{-3}$
- ▶  $\sigma_{He} \propto \nu^{-2}$
- ▶ for  $h\nu > 24.6\text{eV} \rightarrow \sigma_{He} > \sigma_H$
- ▶ HI and OI same threshold
- ▶ KROME database (based on Verner+96)
- ▶ see Tommaso's talk





$$k_{ph} = \frac{4\pi}{h} \int_{E_0}^{\infty} I(E) \frac{\sigma(E)}{E} dE$$

- ▶ What kind of spectra should we expect/use?

# Photoionization

typical spectra



$$k_{ph} = \frac{4\pi}{h} \int_{E_0}^{\infty} I(E) \frac{\sigma(E)}{E} dE$$

- ▶ What kind of spectra should we expect/use?
- ▶ It depends on the environment!

# Photoionization (cont'd)

typical spectra



## Blackbody radiation pervades much of astrophysics

- ▶ The surfaces of “normal” stars emit a spectrum that approximates blackbody radiation
- ▶ the 3-K cosmic microwave background radiation (CMB) exhibits a nearly perfect blackbody spectrum
- ▶ radio spectra from emission nebulae manifest the rising power-law character of blackbody radiation at low frequencies



- ▶ proper BB emission only in optically thick media → photons in perfect thermal equilibrium with particles
- ▶ a unique temperature  $T$  can be defined

## In contrast

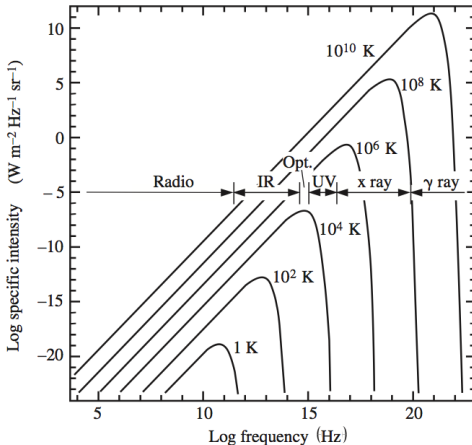
- ▶ star's atmosphere temperature  $\propto$  height
- ▶ gas-photons in equilibrium only in local regions  
→ *Local thermodynamic equilibrium (LTE)*
- ▶ emission: BB spectrum of temperature  $T$  only in small region.
- ▶ the spectrum in another nearby region will approximate the blackbody form at the somewhat different temperature of that region.

# Typical spectra

black-body intensity (eV/s/cm<sup>2</sup>/Hz/sr)



$$B(\nu, T) = \frac{2h\nu^3}{c^2} \frac{1}{(e^{\frac{h\nu}{kT}} - 1)} \implies B(E, T) = \frac{2E^3}{h^2c^2} \frac{1}{(e^{\frac{E}{kT}} - 1)}$$



# Useful quantities

flux and number of photons per seconds



The energy flux density (considering a hemisphere)

$$\mathcal{F}_\nu = \pi B(\nu, T) \rightarrow \mathcal{F} = \pi \int_0^\infty B(\nu, T) d\nu = \sigma_{SB} T^4 \quad [\text{eV s}^{-1} \text{cm}^{-2}]$$

*Under the assumption that a star emits a BB radiation, the luminosity of a star can be expressed as*

$$L_\nu = 4\pi R_\star^2 \mathcal{F}_\nu \rightarrow L_\star = 4\pi R_\star^2 \mathcal{F} = 4\pi R_\star^2 \sigma_{SB} T^4$$

## Number of photons emitted

$$N_{ph}[1/s] = \int_0^\infty \frac{L_\nu}{h\nu} d\nu$$
$$N_{ph}[1/s] = 4\pi R_\star^2 \int_0^\infty \frac{\pi B(\nu, T)}{h\nu} d\nu$$



# Useful quantities

spectral flux density [ $\text{eV s}^{-1} \text{cm}^{-2}$ ]

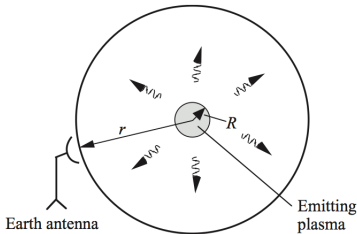


$$S(\nu, T) = \frac{1}{4\pi r^2} \int_0^\infty L_\nu(\nu) d\nu$$

$\Rightarrow$  in terms of luminosity

$$S(\nu, T) = \pi \frac{R_*^2}{r^2} \int_0^\infty B(\nu, T) d\nu$$

$\Rightarrow$  in terms of BB radiation



If you know at which  $T$  emits your star, the radius  $R_*$ , and the distance from the observer (or a decay region  $r$ ) you can evaluate the emissivity in units of  $\text{eV s}^{-1} \text{cm}^{-2}$ .

(see the afternoon exercise)



Typical spectra adopted in modelling:

- ▶ stellar sources → black-body at  $T_{\text{eff}}$
- ▶ quasars sources  $\propto (\nu/\nu_H)^{-\alpha}$  with different power-law exponent
  - ▶  $\alpha = 1.0$  for FUV sources
  - ▶  $\alpha = 1.5$  for X-rays sources
- ▶ interstellar radiation field → Draine flux
  - ▶  $N_{\text{ISRF}} = 8.530 \times 10^{-5} \lambda^{-1} - 1.376 \times 10^{-1} \lambda^{-2} + 5.495 \times 10^1 \lambda^{-3}$

Often the flux is normalized to the value at the Lyman limit

$$10^{-21} J_{21} \frac{B(\nu, T_{\text{eff}})}{B(\nu_H, T_{\text{eff}})} \text{ erg cm}^{-2} \text{ Hz}^{-1} \text{ s}^{-1} \text{ sr}^{-1}$$

It is also quite common to find photodissociation rates already integrated over the Draine flux and depending on the extinction coefficient  $A_\nu$ .

- ▶ integrated from the threshold  $\nu_i$  to the Lyman-limit ( $\nu_H = 13.6 \text{ eV}$ )
- ▶ mostly for diffuse interstellar medium applications
- ▶  $k_{\text{pd}} = a \exp[-bA_\nu]$  (see photodissociation slides)

# Main processes induced by radiation (2)

photodissociation



Molecules can both photoionize or/and photodissociate in the presence of radiation



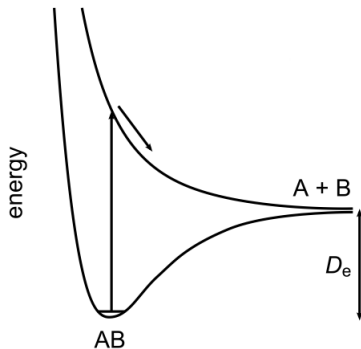
- ▶ H<sub>2</sub> photoionization potential is 15.42 eV
- ▶ mostly in HII regions



- ▶ photodissociation is a complex process
- ▶ it can proceed through three different paths

# Main processes induced by radiation (2)

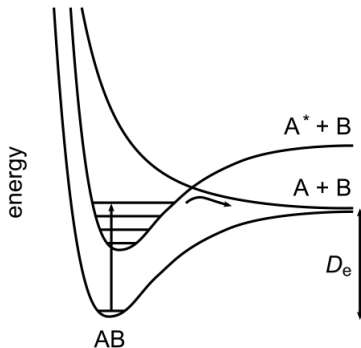
direct photodissociation



- ▶ electronic transitions (**discussed by Daniele**)
- ▶ a molecule absorbs a photon into an excited electronic state that is repulsive with respect to the nuclear coordinate.
- ▶ spontaneous emission back to the ground state is a slow process ( $A \sim 10^9 \text{ s}^{-1}$ )
- ▶ dissociation times of  $10^{13} \text{ s}^{-1}$
- ▶ virtually all of the absorptions lead to dissociation of the molecule.

# Main processes induced by radiation (2)

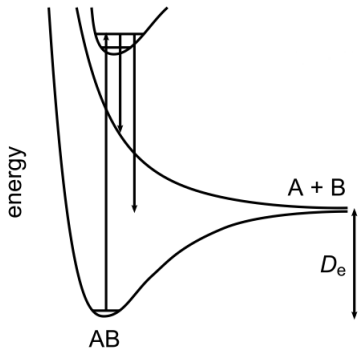
predissociation



- ▶ initial absorption occurs into a bound excited electronic state
- ▶ non-radiatively interaction with a nearby repulsive electronic state
- ▶ e.g. spin-orbit coupling between states of different spin multiplicity (pure quantum mechanics → let's avoid details)

# Main processes induced by radiation (2)

spontaneous radiative dissociation



- ▶ if the excited bound states are not predissociated
- ▶ emission of photons into the continuum of a lower-lying repulsive state
- ▶ emission into the vibrational continuum of the ground electronic state

# Main processes induced by radiation (2)

rates



$$1. k_{ph} = \frac{4\pi}{h} \int_{E_0}^{\infty} \frac{I(E)\sigma(E)}{E} dE$$

$$2. k_{ph} = \sum_{lines} \frac{\pi e^2}{m_e c^2} \lambda_{line}^2 \eta_{line} I(\lambda_{line})$$

$$3. k_{ph} = a \exp[-bA_\nu]$$

- ▶ continuum case (direct dissociation)
- ▶ discrete case (predissociation & spontaneous radiative dissociation)
- ▶ already integrated, including optical depth

- ▶ molecules like OH, H<sub>2</sub>O, CH, CH<sub>2</sub>, H<sub>2</sub><sup>+</sup> → direct photodiss. (1)
- ▶ CO, H<sub>2</sub>, N<sub>2</sub> → discrete formula (2)
- ▶ some of the molecules can go through both processes depending on the radiation:
  - ▶ Direct photodissociation in hot radiation fields
  - ▶ Predissociation in cool radiation fields

Note: small  $A_\nu$  optically thin medium, high  $A_\nu$  optically thick medium. It works as an opacity term (see next slides).

# Main processes induced by radiation (2)

## H<sub>2</sub> photodissociation



Why H<sub>2</sub> is particular?

- ▶ dissociation energy at 4.48 eV
- ▶ homonuclear molecule/no dipole-moment

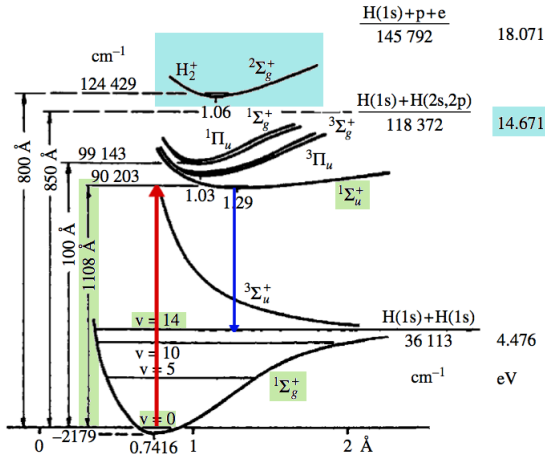
H<sub>2</sub> photodissociation can occur by

1. Direct excitation to the vibrational continuum of the ground electronic state
  - ▶ strongly forbidden
  - ▶ proceeds at a negligible rate
2. Excitation to the vibrational continuum of an excited electronic state of H<sub>2</sub>
  - ▶ threshold of 14.16 eV for ortho-hydrogen
  - ▶ threshold of 14.68 eV for para-hydrogen
  - ▶ restricted to HII regions
3. Two-step photodissociation → "Solomon process" (10% of the emission goes into the continuum, Stecher and Williams, 1967)
4. absorptions into the Lyman and Werner bands at 912–1100 Å



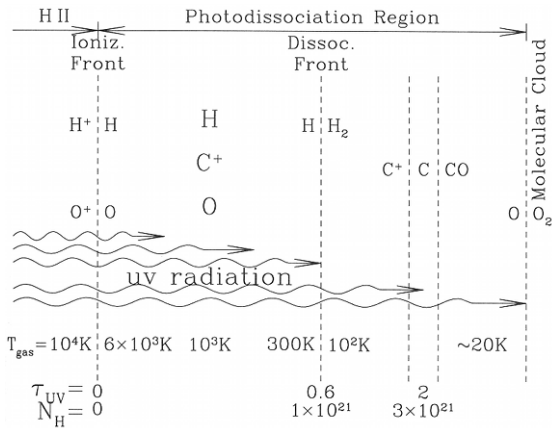
# Main processes induced by radiation (2)

H<sub>2</sub> Solomon process via Lyman band B→X



# Optical depth

understanding from a PDR



**Figure 31.2** Structure of a PDR at the interface between an H II region and a dense molecular cloud.



optically thin

$$k_{phi} = \frac{4\pi}{h} \int_{E_0}^{\infty} \frac{I(E)\sigma(E)}{E} dE$$

optically thick

$$k_{ph} = \frac{4\pi}{h} \int_{E_0}^{\infty} \frac{I(E)\sigma(E)}{E} e^{-\tau(E,n)} dE$$

▶ optical depth

▶  $\tau(E, n) = \sum_i \sigma_i N_i$

▶  $\sigma_i \rightarrow$  cross-section

▶  $N_i \rightarrow$  column density (see Tommaso's talk)

limiting cases

▶  $\lim_{\tau \rightarrow 0} e^{-\tau} = 1 \implies$  medium optically thin

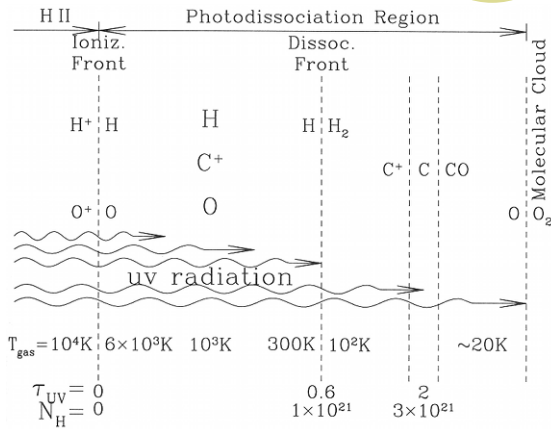
▶  $\lim_{\tau \rightarrow \infty} e^{-\tau} = 0 \implies$  medium optically thick

# Optical depth

understanding from a PDR



- ▶  $\lim_{\tau \rightarrow 0} e^{-\tau} = 1$
- ▶  $\lim_{\tau \rightarrow \infty} e^{-\tau} = 0$
- ▶ HII regions optically thin
- ▶ HI region optically thick



**Figure 31.2** Structure of a PDR at the interface between an HII region and a dense molecular cloud.

# Main processes induced by radiation (3)

photoheating



The photodissociation and photoionization induced by FUV radiation generate an excess of energy which can go into heating ( $h\nu - E_0$ ).

Photoheating is mainly caused by

- ▶ atoms photoionizations in HII regions ( $h\nu > 13.6$  eV)
- ▶ photo-ionization of large molecules and small dust grains in HI regions ( $h\nu < 13.6$  eV)
- ▶ molecules photodissociation in molecular regions

$$H_{ph}[\text{erg s}^{-1}] = \frac{4\pi}{h} \int_{E_0}^{\infty} \frac{I(E)\sigma(E)}{E} (E - E_0)\eta(E)e^{-\tau} dE \quad (1)$$

$\eta(E)$  is an efficiency factor that determines the amount of energy released into the gas. The effective photoheating is

$$\Gamma_{ph} = H_{ph}n_X \quad (2)$$

in  $\text{erg s}^{-1} \text{ cm}^{-3}$ .

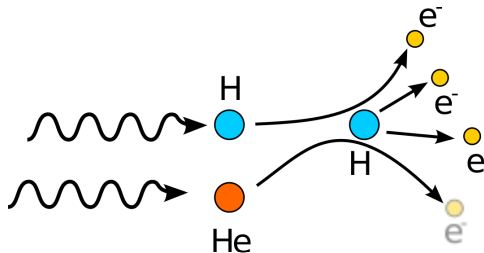
# Main processes induced by radiation (4)

X-rays photons



X-rays photons (2-10 keV)

- ▶ a typical X-ray photon is far more likely to be absorbed by HeI rather than HI.
- ▶ generate energetic photoelectrons which cause secondary ionizations
- ▶ the ejected photoelectron, however, will ionize many more HI atoms than HeI, as HI is more abundant



Courtesy of T. Grassi

Shull+1985, Wolfire+1995, Ricotti+2002, Ricotti+2004, 2005, Furlanetto+2010

# Main processes induced by radiation (4)

X-rays photons



$$\zeta_{tot}^i = \zeta_p^i + \sum_{j=\text{H,He}} \frac{n_j}{n_i} \zeta_p^j \langle \phi^i \rangle$$

► total primary + secondary ionization rate

# Main processes induced by radiation (4)

X-rays photons



$$\zeta_{tot}^i = \zeta_p^i + \sum_{j=H,He} \frac{n_j}{n_i} \zeta_p^j \langle \phi^i \rangle$$

- ▶ total primary + secondary ionization rate
- ▶ primary photoionization rate



# Main processes induced by radiation (4)

X-rays photons



$$\zeta_{tot}^i = \zeta_p^i + \sum_{j=H,He} \frac{n_j}{n_i} \zeta_p^j \langle \phi^j \rangle$$

- ▶ total primary + secondary ionization rate
- ▶ primary photoionization rate
- ▶ number of secondary ionization per primary ionization

# Main processes induced by radiation (4)

X-rays photons



27

$$\zeta_{tot}^i = \zeta_p^i + \sum_{j=H,He} \frac{n_j}{n_i} \zeta_p^j \langle \phi^j \rangle$$

- ▶ total primary + secondary ionization rate
- ▶ primary photoionization rate
- ▶ number of secondary ionization per primary ionization
- ▶  $\phi^H(E, x_e) = \left( \frac{E}{13.6 \text{ eV}} - 1 \right) 0.3908(1 - x_e^{0.4092})^{1.7592}$

# Main processes induced by radiation (4)

X-rays photons



27

$$\zeta_{tot}^i = \zeta_p^i + \sum_{j=H,He} \frac{n_j}{n_i} \zeta_p^j \langle \phi^i \rangle$$

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$$\zeta_p^H(E, x_e) = \left( \frac{E}{13.6 \text{ eV}} - 1 \right) 0.3908 (1 - x_e^{0.4092})^{1.7592}$$

$$\zeta_p^{He}(E, x_e) = \left( \frac{E}{24.6 \text{ eV}} - 1 \right) 0.0554 (1 - x_e^{0.4614})^{1.666}$$

# Main processes induced by radiation (4)

X-rays photons



27

$$\zeta_{tot}^i = \zeta_p^i + \sum_{j=H,He} \frac{n_j}{n_i} \zeta_p^j \langle \phi^i \rangle$$

- ▶ total primary + secondary ionization rate
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$$\zeta_p^{He}(E, x_e) = \left( \frac{E}{24.6 \text{ eV}} - 1 \right) 0.0554 (1 - x_e^{0.4614})^{1.666}$$

$$\langle \phi^i \rangle = \frac{\int I_X(E) \phi^i(E, x_e) dE}{\int I_X(E) dE}$$

# Main processes induced by radiation (4)

X-rays photons



$$\zeta_{tot}^i = \zeta_p^i + \sum_{j=H,He} \frac{n_j}{n_i} \zeta_p^j \langle \phi^i \rangle$$

- ▶ total primary + secondary ionization rate
- ▶ primary photoionization rate
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$$\zeta_p^H(E, x_e) = \left( \frac{E}{13.6 \text{ eV}} - 1 \right) 0.3908 (1 - x_e^{0.4092})^{1.7592}$$

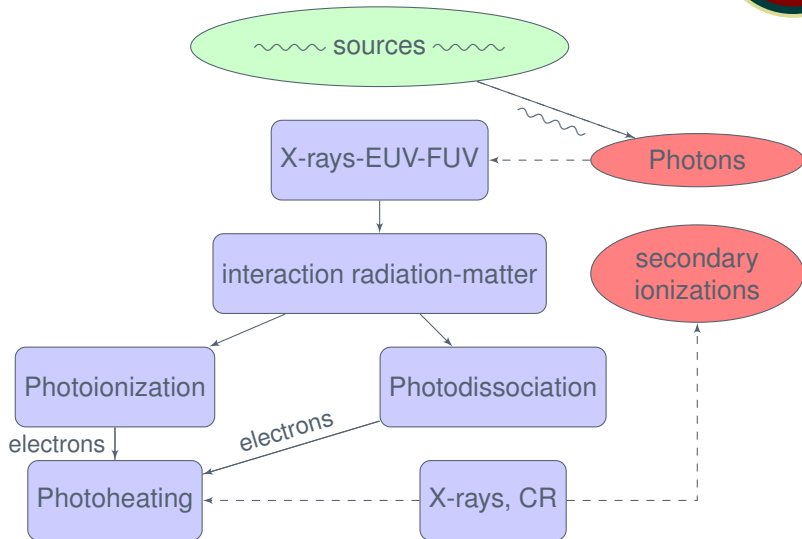
$$\zeta_p^{He}(E, x_e) = \left( \frac{E}{24.6 \text{ eV}} - 1 \right) 0.0554 (1 - x_e^{0.4614})^{1.666}$$

$$\langle \phi^i \rangle = \frac{\int I_X(E) \phi^i(E, x_e) dE}{\int I_X(E) dE}$$

- ▶ → see Latif's talk for an application

# Interaction radiation-matter

a schematic summary





## References:

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[www.kromepackage.org](http://www.kromepackage.org)

Thank you for your attention!

